

# The Oil Report July 2011

Oil the News that's Fit to Print!

What's the best oil to use? Everyone has an opinion, and the truth is, there isn't one best answer. We've written about this very issue here. Got some time on your hands? Feel free to click around on our vast trove (er...maybe only somewhat vast) of other oil-related articles.

## **Phantom Fuel**

by Amanda Callahan

When I first started at Blackstone, one of the contaminants that intrigued me the most was fuel. I guess I don't know why finding fuel in people's oil surprised me. Maybe I thought fuel and oil were separated by a giant wall somewhere in the engine. Or maybe I really didn't understand engines very well to begin with and that only served to fuel (ha!) my interest in the contaminant.

We test for fuel using the Cleveland Open Cup method. Basically, we record the temperature at which the vapors from the oil ignite. All oils have a specification for what the flashpoint should be. When it's lower than that, it's because a contaminant is present. About 98% of the time, that contaminant is fuel (sometimes a solvent or refrigerant will lower a flashpoint, but that's rare in gas or diesel engines). Basically, the lower the flashpoint, the more fuel you've got. We can accurately measure fuel down to 0.5%, so that's the lowest fuel measurement you'll see on your report. The upper limit of what we can accurately read is 10.0%. If you've got more fuel than 10.0%, you've got bigger problems to worry about than the actual quantity of fuel in the oil.

When the opportunity came up to write an article for the summer newsletter, I readily accepted and already knew I wanted to write about fuel. In fact, I was not just going to write about it—I was going to get to the bottom of it. I was going to discover what causes fuel dilution and what causes fuel to disappear.

#### The Plan

The guinea pig was my trusty 2007 Kia Optima (2.4L, 4-cylinder). I use my Kia mostly as a daily driver, traveling about seven miles to and from work each way. I love to travel and occasionally I get in a trip to Wisconsin or Iowa. By the time I started my quest to debunk fuel, I'd done a few samples with my Kia and only a trace of fuel had ever turned up, so I didn't have any known fuel system problems to contend with.

I decided to take the highway route home every day to ensure that every day I would cook out any extra fuel that was present in my oil. My 40-minute drive consisted of some city streets with a few stoplights at the beginning and end of my trip, and mostly sustained highway speeds through the middle of my trip.

#### Start your engines

The first thing I wanted to test was how much fuel entered the oil simply from starting the engine. Many people believe that starting an engine is one of the most taxing and wearproducing events throughout the engine's life. To make that process easier. Like us on engines tend to start slightly rich (more fuel, less air). So, I set out to find out

exactly how much fuel my car dumps into the oil upon startup. After letting the engine sit all night, I took a pre-experiment test sample (to ensure no fuel was present) then I started my engine one, two, and three times, sampling after each event. The results were surprising. So surprising, in fact, that I re-ran this test two more times to make sure my results weren't just an anomaly.

The pre-experiment test sample revealed a flashpoint of 360°F (fuel at <0.5%). Okay, good. No measurable fuel was present, which is exactly what I was hoping for since I'd taken the highway route home the night prior. After one engine start, the flashpoint read 385°F. Wait a minute. That's higher than the presample, so there's definitely still no measurable fuel present. Okay, maybe that was just an anomaly. I

started the engine again (for a total of two engine starts in a row). The flashpoint measured 380°F. That meant the flashpoint was heading in the expected direction (lower flashpoint = more fuel), but still, the flashpoint wasn't low enough to show any significant fuel. After the third start in a row, the flashpoint read 375°F, which was again lower (and heading in the expected direction), but not low enough to show any measurable fuel. So all four of my samples from that morning had the same fuel measurement: <0.5%.

SUS Viscosity @ 210°F	51.7	46-59
cSt Viscosity @ 100°C	7.80	6.0-10.2
Flashpoint in °F	365	>355
Fuel %	<0.5	<2.0

D'oh! No fuel to speak of for the five-minute idle.

I was stumped. I was so certain I'd have some fuel in my oil! So I ran the test again and the same thing happened: no measurable fuel present in any of the samples. I re-ran the same exact test once again with similar results. (I think one definition of insanity is trying the same thing repeatedly while expecting different results, so you can call me crazy).

I decided perhaps my Kia was just very good about keeping fuel out of the oil, so I took my husband's car (a 2005 Saturn Ion Redline with a supercharged 2.0L Ecotec engine) for a day and tried the same test. What'd I get? Same thing: no fuel, with slight fluctuations in the flashpoint.

Then I had an "a-ha!" moment. Maybe the fuel just wasn't getting a chance to seep down past the rings into the oil. I had been sampling immediately after starting, so maybe that's why no fuel showed up. So I ran

the test again, only this time I let my engine sit overnight after the three starts. In the pre-experiment test, the flashpoint measured 370°F, showing <0.5% fuel. I started my engine three times and immediately after the third start, the flashpoint measured 360°F, which was again lower but still not low enough to show any measurable fuel. I let the engine sit all night and sampled before work in the morning. That test revealed a flashpoint of 365°F; still no measurable

SUS Viscosity @ 210°F	52.4	46-59
cSt Viscosity @ 100°C	8.00	6.0-10.2
Flashpoint in °F	335	>355
Fuel %	1.0	<2.0

The ten-minute idle. More fuel, but still not as much as we thought there would be.

fuel. By this point I wanted to find a brick wall to repeatedly press against my forehead in a semi-violent manner. I was frustrated, confused, and worried that I'd have nothing to write about.

I suppose if we wanted to get into semantics, we could talk about the slight differences in flashpoints as showing some fuel, though it takes a 20°F drop in flashpoint (for gasoline engines) to show 1.0% fuel dilution. So that means the 5°F drops in flashpoint I'd noticed likely show just 0.25% fuel.

Does starting cause fuel? Perhaps. I did find slight dips in the flashpoint, though as I've mentioned, 5°F isn't enough to show any serious contamination. Maybe four starts would have given me enough of a drop

in flashpoint to get a decent amount of fuel in the oil, but really, who starts their engine four times in a row? Honestly, who starts their engine even three times in a row on a regular basis? I wanted these to be relatively real-world scenarios, so I couldn't justify four starts in a row, and I figured three was pushing it.

#### Idling

I couldn't get any serious fuel to appear in my oil from starting the engine, so I figured I'd try idling. For a week, I drove the same highway route home, let my engine sit all night, then in the morning I'd take a pre-experiment sample (to confirm no fuel dilution was present to begin with), and then I'd sample after a certain number of minutes of idling. I tried a five-minute idle; no measurable fuel. The next morning, I idled for ten minutes and this time I had more success: 1.0% fuel had accumulated in the oil. You know, for someone trying to get fuel contamination in my oil, 1.0% isn't an impressive amount, but it's more than I'd gotten before. I figured I was on to something with the idling, so I tried 15 minutes the next morning, but much to my dismay only 0.5% fuel turned up. After 20 minutes of idling, still only 0.5% fuel.

Does idling cause fuel dilution? It would seem so, except that there's a cut-off point in there somewhere. This is just hypothetical, but maybe after ten minutes the engine heats up enough that it either starts cooking off the excess fuel dilution or it just stops pumping in extra fuel. I'm not even sure one of those is the answer, but it's the best guess I can come up with.

#### **Shopping for Science**

With my newsletter article deadline quickly approaching, I had to come up with one last-ditch effort to get a bunch of fuel in my oil. Think of this as Custer's Last Stand (except with less bloodshed). For one afternoon, I vowed to do everything "wrong" in order to get as much fuel as possible in my oil. I was going to run a bunch of errands, idle my engine excessively, make frequent starts and short trips. The best way I could think to do this (without just circling around my block several times in one afternoon) was in one, massive shopping trip.

I did about 40 minutes of highway driving Saturday evening then let my Kia sit all night. Sunday after church (we took the other car to ensure the consistency of my results), I went on my scientific shopping trip.

Here's the summary of my trip. I spent a total of about six hours shopping Sunday afternoon. In those five hours, I started my engine seven times, idled at the ATM for about 2 minutes and traveled a grand total of 6.4 miles. The longest drive was from my last stop to home, which was about 2.5 miles. I spent a fair amount of time at each stop in hopes that my engine would stay relatively cool (so as to not burn off fuel). When all was said and done, I left my engine sit overnight and sampled in the morning before work. The flashpoint read 375 °F: <0.5% fuel.

Usually, I'm fairly good at things when I put my mind to it, but when it comes to getting fuel dilution in my engine oil, I failed. Miserably. Then again, if you're going to fail at anything, this is a good thing to fail at.

#### Conclusion

So why couldn't I get any serious fuel dilution? I have a couple of ideas. First, I think my Kia is just too smart. It has an on-board computer that senses things like ambient temperature, engine temperature, and elemental composition of the exhaust gas, and it uses these factors to calculate the exact amount of fuel it needs to operate most efficiently. So my engine never puts in more fuel than it needs, and therefore that fuel doesn't end up in my oil.

Second, I think ambient temperature probably had something to do with it. I did most of my testing in May and June in temperatures were almost always above 70°F. We tend to see more fuel in the winter months,

and I suspect that if I'd done my testing in the winter, I might have had different results. I've heard that on a cold day, fuel from the air/fuel mixture will condense on the cylinder walls almost instantaneously. Those beads of fuel will sit on the cylinder walls for a brief moment until the piston rings scrape that fuel down into the oil. Maybe I'd get more fuel in my oil in winter, but I'm not sure I'm willing to do these experiments in the dead of winter in the name of science. If I do, you'll hear from me again and I'll let you know what I find out.

Does the fact that I couldn't get any fuel in the oil mean that idling, city driving, and frequent starts do NOT cause fuel dilution? We don't think so. In some cases these things can cause fuel contamination, especially in carbureted engines. We saw it with our own eyes several years ago, when an intern did a similar experiment out in the parking lot. He took a sample from his 1978 Ford pickup truck when it was cold, and that oil

had no fuel in it. Then he started the engine and took another sample right away. Presto! Fuel contamination at 1.3%. So start-up can indeed cause fuel to enter into the oil, but newer engines may be better at avoiding excessive contamination. Since fuel often comes and goes, we still believe operational factors are likely sources for fuel dilution, though perhaps injector problems are responsible for more of the fuel we see in new engines than we originally thought.

SUS Viscosity @ 210°F	73.7	64-73
cSt Viscosity @ 100°C	13.93	11.3-14.0
Flashpoint in °F	350	<b>&gt;37</b> 5
Fuel %	1.3	<2.0

Fuel did show up after just one start in this '78 Ford. Clearly the engine type has a lot to do with whether and how much fuel you find in your oil.

Now here's the big question: if fuel is present at 2.0% in your sample, does that mean you have a problem? Not necessarily. Just because my Kia didn't produce fuel dilution doesn't mean your Honda, Ford, GM, Volvo, or other engine won't. Every engine operates a little differently and uses a different calculation to figure out how much fuel to spray into the cylinders. Some engines tend to run a little richer than others. Of course, if your engine doesn't have an on-board computer, you won't have the opportunity to benefit from its fuel-reducing powers, so fuel may be more prevalent in your samples. It should be noted that I didn't have the opportunity to test a carbureted engine or a diesel engine, which would almost certainly render different results. So I can't say what's normal for those types of engines.

So how do you know if fuel is a problem? There isn't a one-size-fits-all answer. Whether or not fuel is a problem depends on your circumstances. Some engines will always see some fuel dilution because of the operation they see or the type of engine they are. Turbo and supercharged engines, for example, have higher compression, which means more blow-by, and we sometimes see that raw fuel blowing past the rings. In small amounts, that can be fine. In larger amounts, it's probably not.

If you start to notice that fuel is increasing wear or diluting the additives in your oil, that can be a sign that fuel's a problem. Fuel dilution, when it's present for a long enough time, can soften the lead Babbitt on some bearings. If you notice increased bearing wear and lingering fuel dilution, it may be time to get the fuel issue taken care of. Finally, if you notice your engine is "making oil" (the oil level seems to be rising on the dipstick), you might have a fuel system problem. I can tell you this: if you have a Kia 2.4L 4-cylinder engine and you find fuel at more than 1.0%, you may have a problem. You might also have an e-mail in your inbox from me asking you for pointers.



### Report of the Month

This Subaru was burning a quart of oil every 350 miles. That's like a continual oil change, yet bearing wear is still way too high. Where did it all go wrong for this engine? Take a look at the data, then check the caption below to see what happened.

To learn more about where the elements are coming from, click here.

	MI/HR on Oil		UNIT/			UNIVERSAL
	MI/HR on Unit		LOCATION AVERAGES			AVERAGES
	Sample Date	01/20/11				
	ALUMINUM	3	3			3
	CHROME	1	1			1
MILLION	IRON	11	11			8
	COPPER	28	28			5
	LEAD	17	17			4
PER	TIN	2	2			1
	MO LYBDENUM	237	237			68
PARTS	NICKEL	0	0			0
Z	POTASSIUM	0	0			2
	BORON	67	67			46
EN	SILICON	5	5			12
ELEMENTS	SODIUM	3	3			26
岀	CALCIUM	1842	1842			2202
	MAGNESIUM	5	5			124
	PHOSPHORUS	818	818			708
	ZINC	942	942			854
	BARIUM	0	0			0

Values Should Be\*

PROPERTIES	SUS Viscosity @210°F	71.9	66-75		
	cSt Viscosity @ 100°C	13.45	11.9-14.5		
	Flashpoint in °F	345	>385		
	Fuel %	2.0	<1.0		
	Antifreeze %	0.0	0.0		
	Water %	0.0	0.0		
	Insolubles %	0.3	0.7		
	TBN				
	TAN				
	ISO Code				

\*THIS COLUMN APPLIES ONLY TO THE CURRENT SAMPLE

The owner writes: After receiving your report, I decided to follow through with an overhaul. When the engine was opened up, the cylinder walls and piston skirts were scored very bad. The crank could file down your fingernails just moving them back and forth across the journal. Apparently the head had been off before and reinstalled with some aftermarket gasket that might have contributed to the problem. There was a lot of sludge in the pan, as well as some shavings. The engine allegedy has only 47K on it. I wonder if it had been overheated somewhere along the line.